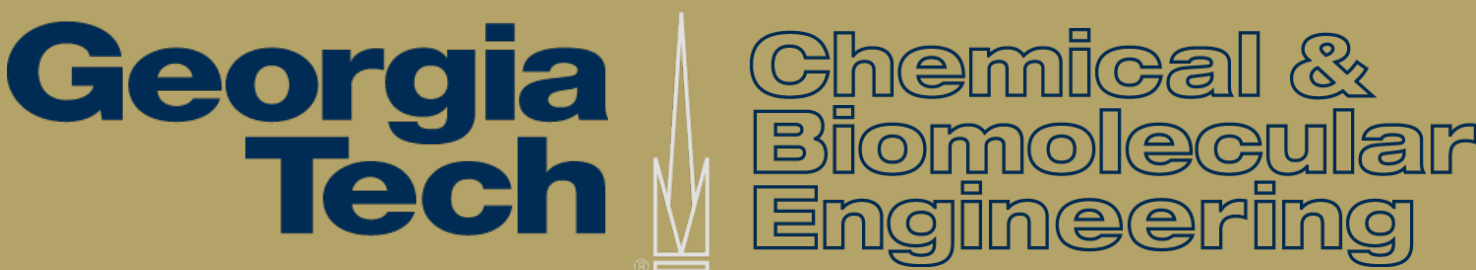


Decoupling Mechanical Deformation in Stack-Pressured Aqueous Zinc Batteries

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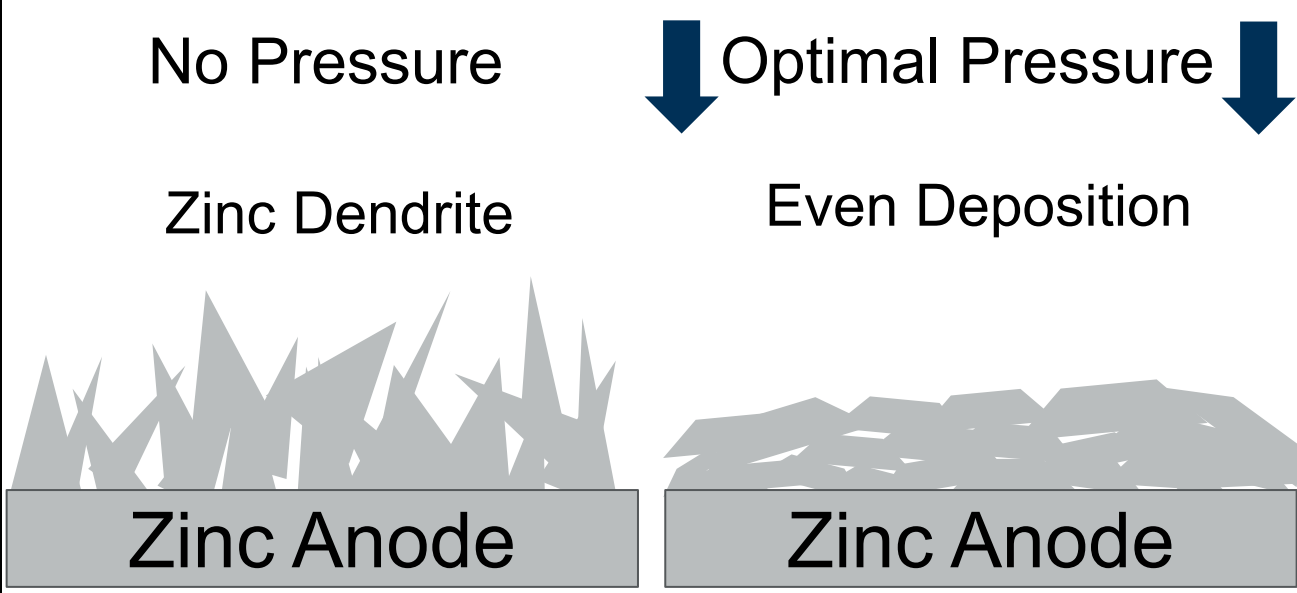
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Topic: Zinc & Lead Batteries | Vertical: Materials and System Innovation | Strategic Pathways: Optimize | Project Start and Finish: FY25 – FY28 | Storage Innovations 2030

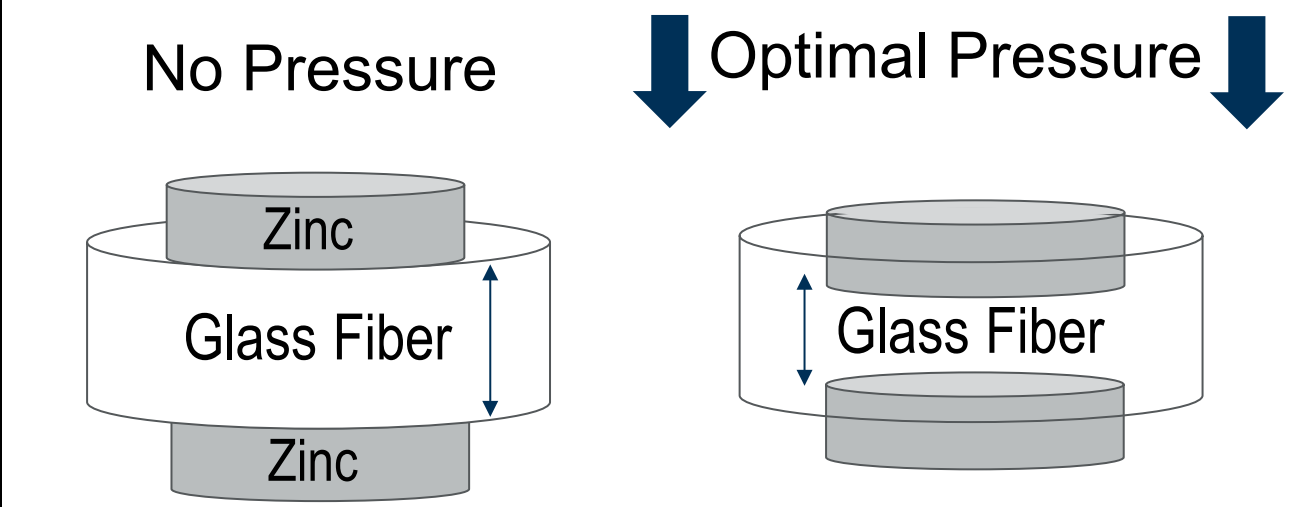


Problem

- **Aqueous zinc (Zn) batteries** offer a low-cost, safe, and sustainable alternative to Li-ion for large-scale energy storage¹
- Frequently fail due to Zn dendrite growth, gas evolution, and interfacial instability
- Researchers have pursued electrolyte and anode modifications to suppress dendrites, these approaches are complex, costly, and difficult to scale²
- **Stack pressure** offers a simple, low-cost physical strategy to promote uniform Zn growth and cycling stability³



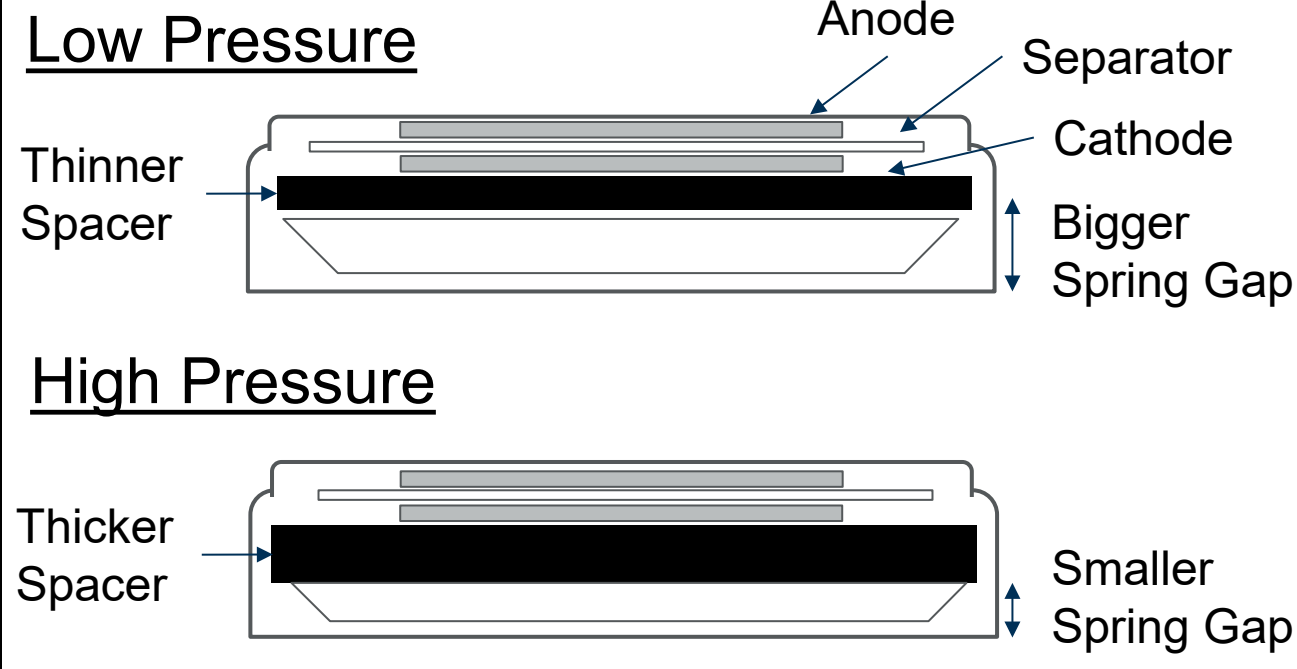
- Prior studies have overlooked mechanical deformation of cell components under applied pressure, particularly the separator



- In this study, we used an incompressible polycarbonate separator to decouple mechanical deformation from stack pressure, isolating pressure's intrinsic effects
- When utilizing stack pressure, **cell components become a large factor** in determining the electrochemical performance

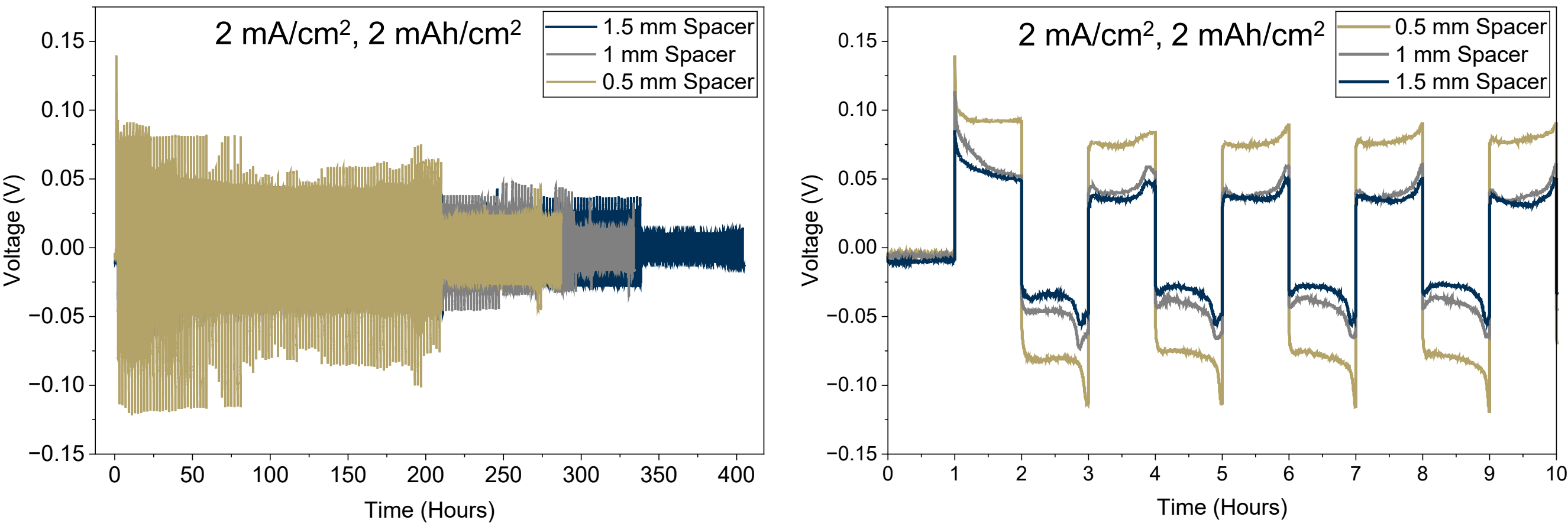
Approach

- Zn Symmetric Cell
 - CR 2032 Coin Cell
 - Zinc Foil Electrode (10 mm Diameter, 0.25 mm Thickness)
 - Stainless Steel Spacer
 - Whatman Glass Fiber B
 - Electrolyte: 2 M ZnSO₄
 - Polycarbonate Separator (0.4 μm Pore Size)
 - Electrolyte: 1.5 m ZnSO₄ + 15 m LiTFSI
- Hookes Law: Force ∝ Displacement



Results

1. Electrochemical Performance with Compressible Glass Fiber



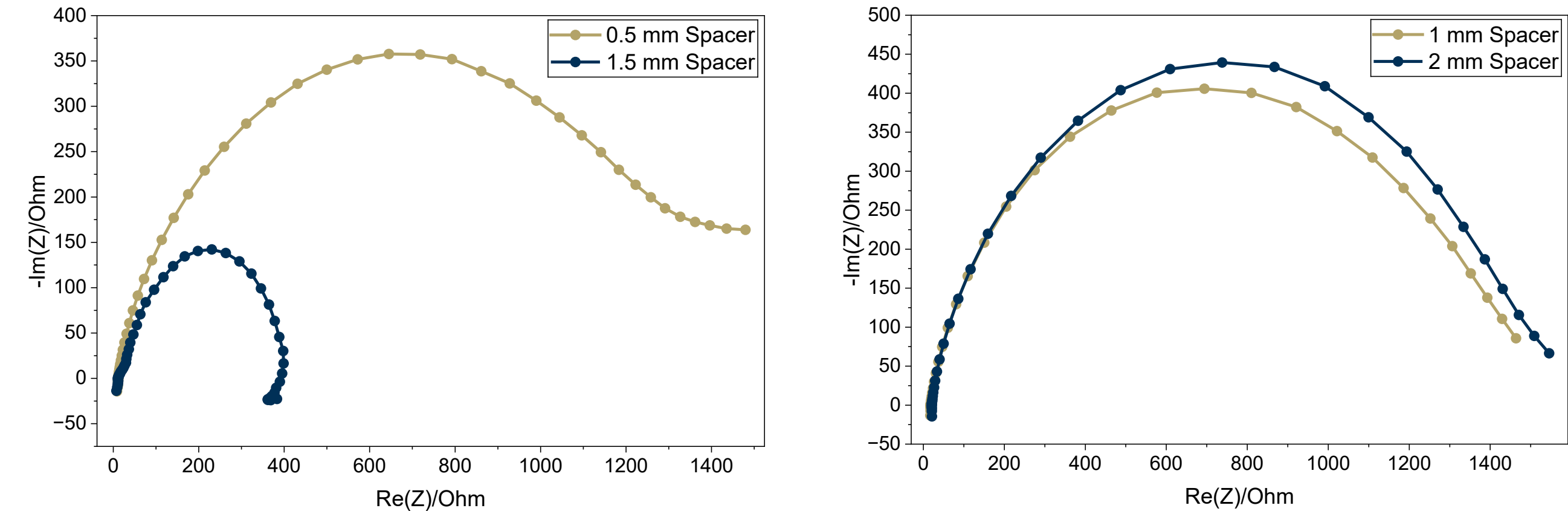
- The cycling performance of zinc symmetric cells using a compressible glass fiber separator
- Increasing stack pressure significantly improves cycle life while reducing overpotential
- However, at higher pressures, separator compression reduces the distance required for dendrites to propagate and short the cell, which limits the beneficial effect of pressure

2. Quantifying Changes in Porosity, Tortuosity, and Electrolyte Wetting in Compressible Glass Fiber

	Low Pressure (0.5 mm Spacer)	High Pressure (1.5 mm Spacer)
Dry Weight	28.9 mg	28.6 mg
Wet Weight	172.9 mg	172.6 mg
After Resting for 48 hours	169.7 mg	135.9 mg
% Electrolyte Lost	~ 2%	~ 21%

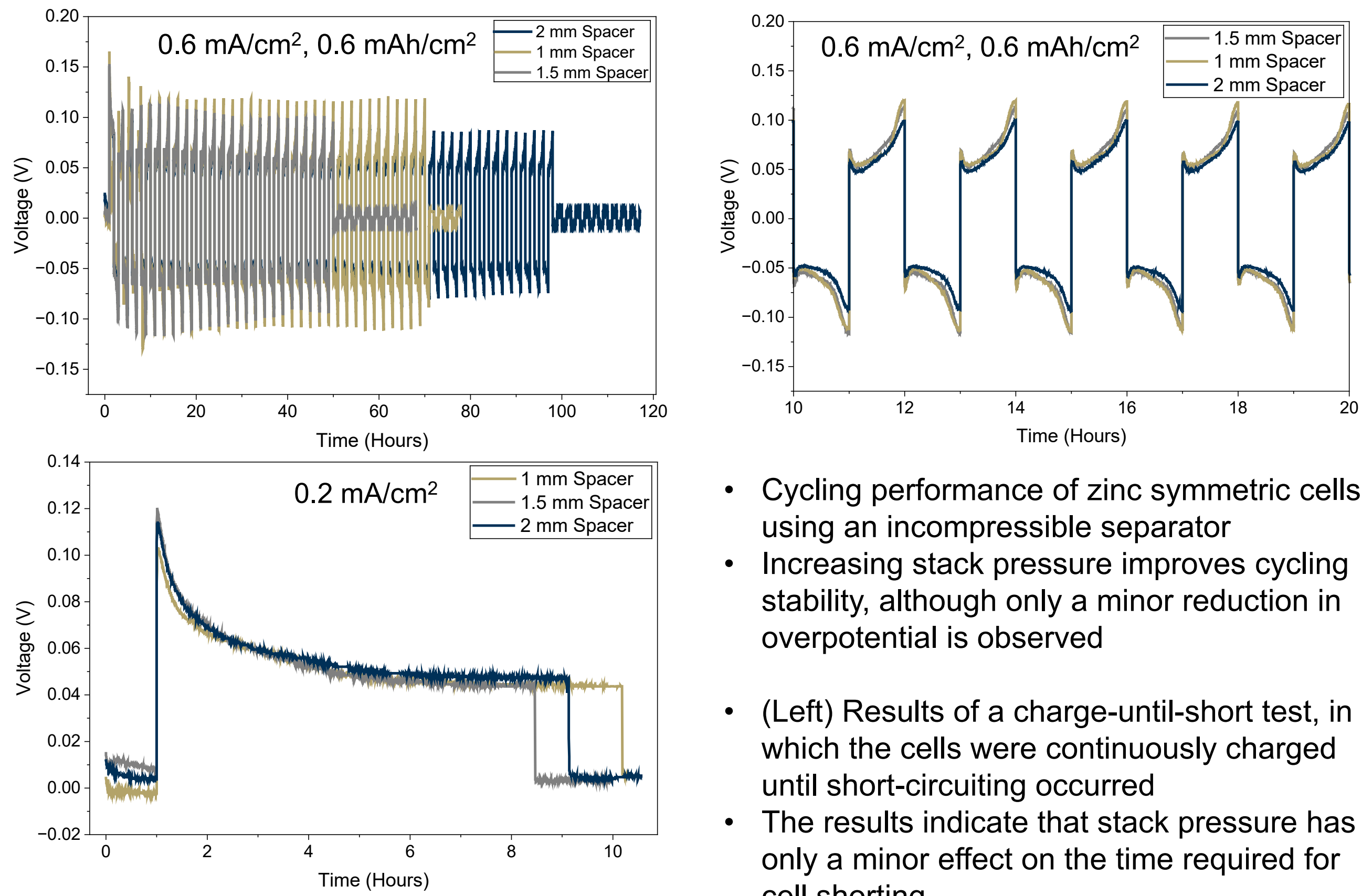
- (Left) Glass fiber separators after compression under low pressure (0.5 mm spacer) and high pressure (1.5 mm spacer) conditions
- Under high pressure, a visible indentation from the zinc electrode is observed along with noticeable dry regions within the separator
- (Right) electrolyte depletion based on dry and wet separator weights
- Wet weight = dry weight + 120 μL of 2 M ZnSO₄, assuming a density of 1.2 mg μL⁻¹

Electrochemical Impedance Spectroscopy (EIS)



- EIS spectra of glass fiber (left) and polycarbonate (right) separators under low- and high-pressure conditions after a 1 h rest period before cycling (1 MHz–100 mHz, 10 mV amplitude)
- The high-pressure glass fiber separator exhibited a reduced semicircle + inward low-frequency tail
- In contrast, the incompressible polycarbonate separator showed only a minor change in resistance with increasing stack pressure

3. Electrochemical Performance with Incompressible Polycarbonate



- Cycling performance of zinc symmetric cells using an incompressible separator
- Increasing stack pressure improves cycling stability, although only a minor reduction in overpotential is observed

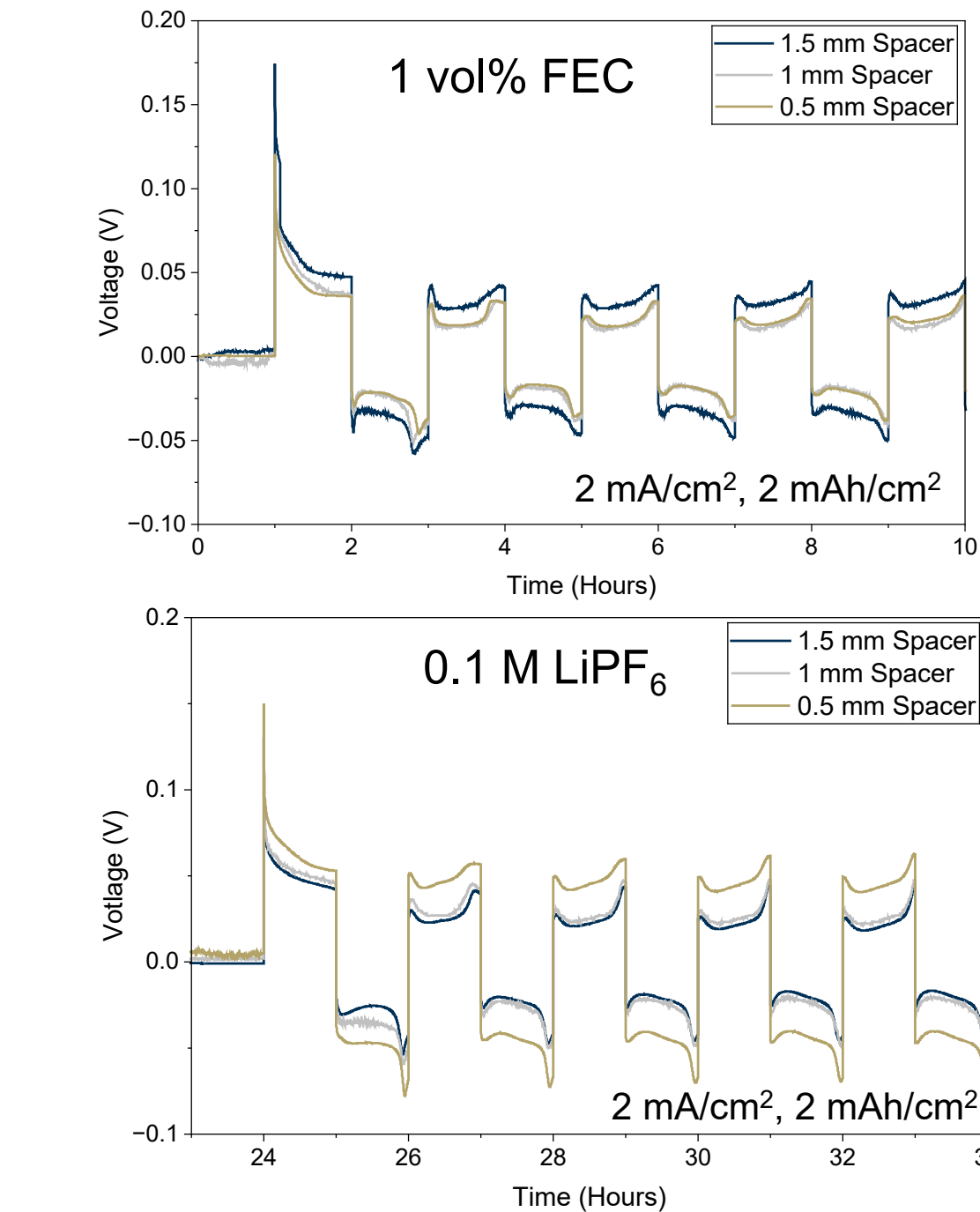
- (Left) Results of a charge-until-short test, in which the cells were continuously charged until short-circuiting occurred
- The results indicate that stack pressure has only a minor effect on the time required for cell shorting

Conclusions & Impact

- Applying stack pressure with a compressible glass fiber separator **increased cycle life by ~125 hours**
- Stack pressure reduced overpotential, improved electrode–separator contact, and suppressed dendrite growth
- Postmortem analysis revealed separator deformation and electrolyte depletion
- EIS spectra of glass fiber separators showed significant changes in porosity and tortuosity. In contrast, incompressible polycarbonate separators exhibited only minor resistance changes
- Using an incompressible separator revealed an optimum pressure regime
 - **Low stack pressure** creates interfacial voids that increase dendrite propagation distance
 - **Medium stack pressure** reduces void formation but provides limited dendrite suppression
 - **High stack pressure** eliminates interfacial voids and promotes dendrite suppression
- Minimal overpotential changes suggest that separator compressibility strongly influences interfacial behavior
- The dendrite suppression mechanism is primarily effective during continuous plating/stripping rather than plating alone

Future Directions

- In situ **x-ray micro-computed tomography** (Micro CT)
 - Probe morphological evolution of battery electrode and components
- Stack pressure effects on **electrolyte additive-induced SEI**



References

1. Kharel, D.; Quilly, C. D.; Bezsonov, I. I.; Wright, C. N.; Lambert, T. N.; Cheng, Y.-T. Effect of Externally Applied Pressure on Rechargeable Alkaline Zinc Batteries at Limited Depth of Discharge. *J. Mater. Chem. A* 2026, Advance Article. <https://doi.org/10.1039/D5TA08160B>
2. Zhang, F.; Cai, Z.; Liu, R.; Sun, Y.; Pan, H. Guiding Uniform Zn Electrodeposition through Regulating Pressure for Stable Aqueous Zn Batteries. *J. Mater. Chem. A* 2024, 12, 10341–10348. <https://doi.org/10.1039/D3TA07523K>
3. Liu, Y.; Yang, Y.; Huang, H.; Wang, Z.; Dong, L.; Zhang, F.; Kang, Y.; Wang, J.-J.; Wang, S.; Liu, H.; Chen, H. Tip-Localized Stack Pressure Field Distribution towards Inactive Zn Loss Reduction for Reversible Zinc Metal Batteries. *Energy Storage Mater.* 2025, 75, 104060. <https://doi.org/10.1016/j.ensm.2024.104060>

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