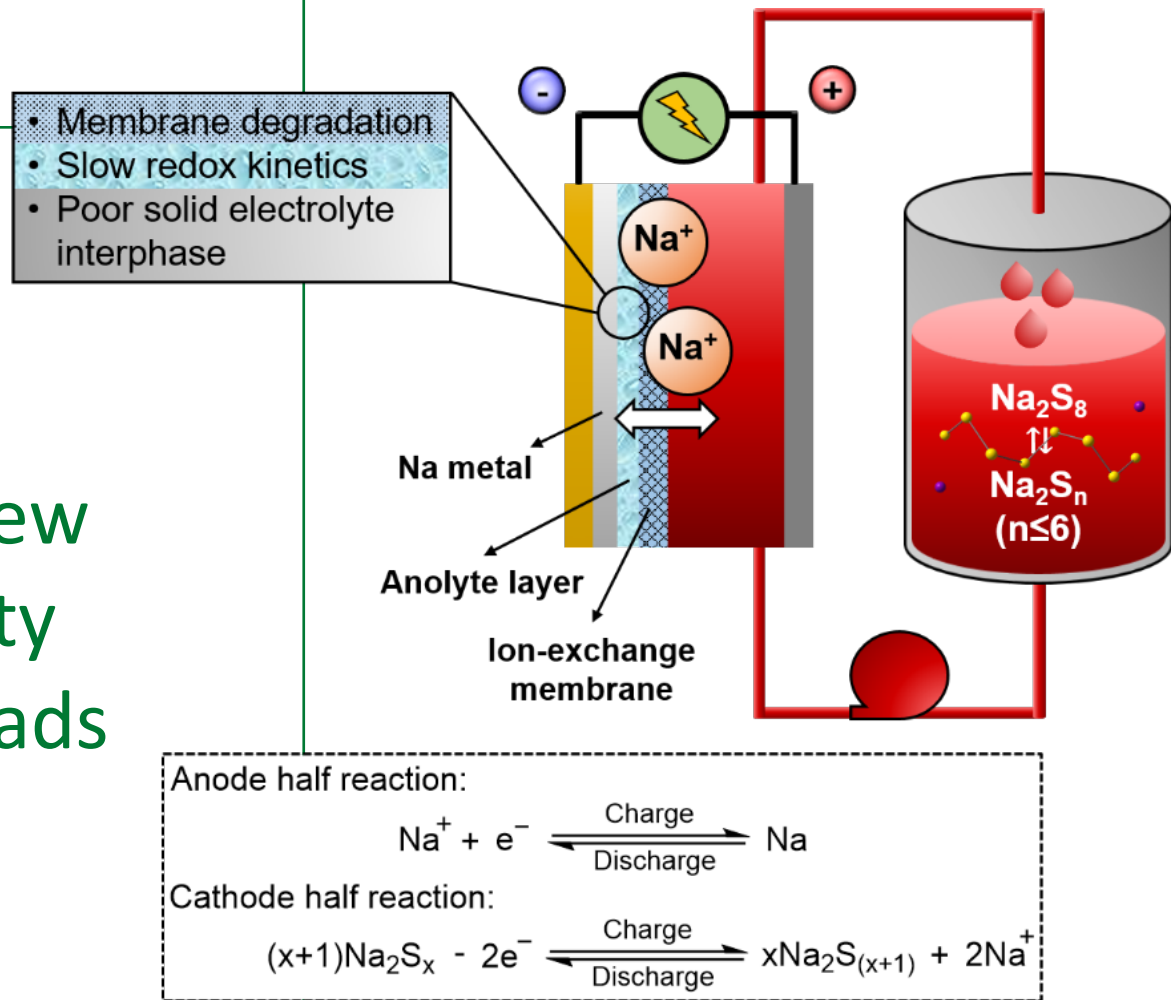


PROJECT PROFILE

Topic: Long-duration energy storage – Flow batteries | Vertical: Materials and Systems Innovation

Strategic Pathway: Grow the grid through new storage technologies; Optimize grid reliability through durable storage; Stabilize critical loads through dispatchable backup storage

Project start and finish: FY25-FY26

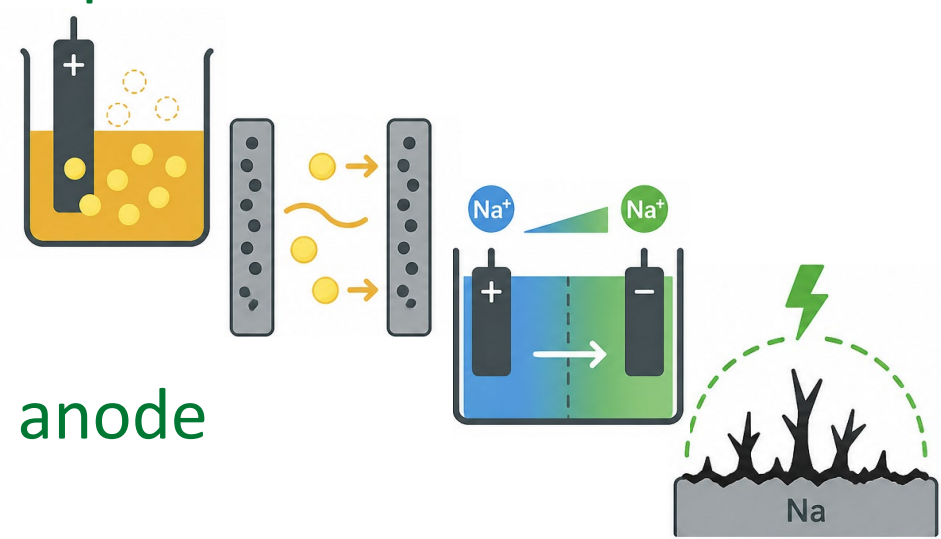


PROBLEM

Background: Grid-scale storage needs technologies that are affordable, reliable, safe, and scalable. Vanadium flow batteries depend on >\$300 /m² Nafion, exceeding 40 % of system cost. Vanadium availability is restricted domestically. This keeps LCOS well above the 5 ¢/kWh-cycle target.

Sodium-polysulfide flow batteries are attractive because they use earth-abundant active materials and decouple energy from power, but performance remains constrained by four coupled losses:

- incomplete polysulfide utilization at the cathode
- species crossover through the separator
- transport imbalance between two tanks
- sluggish/unstable Na⁰/Na⁺ exchange at the metal anode



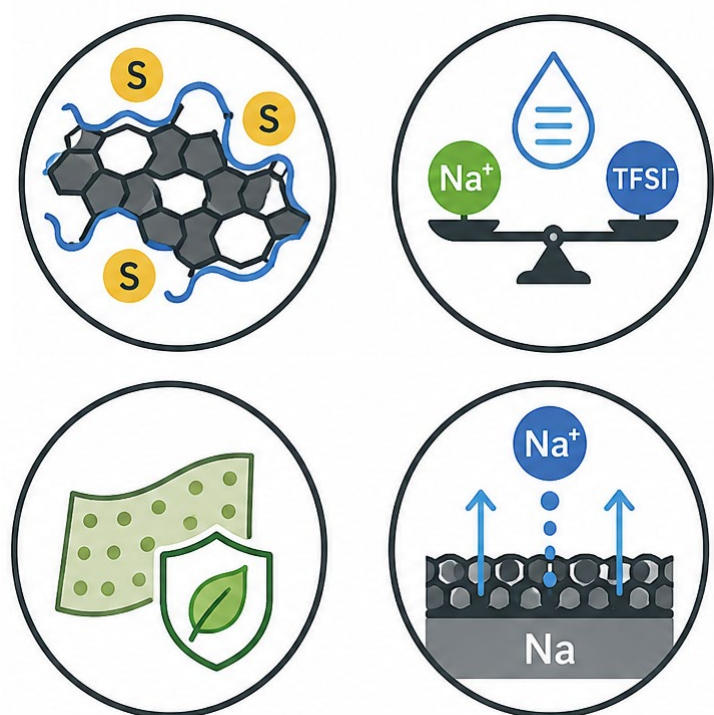
Solving only one of these bottlenecks is not enough; the full stack must be engineered as an ion-transport problem.

APPROACH

Hypothesis: Na–S flow batteries can be improved by controlling transport at each interface, not by optimizing a single component in isolation.

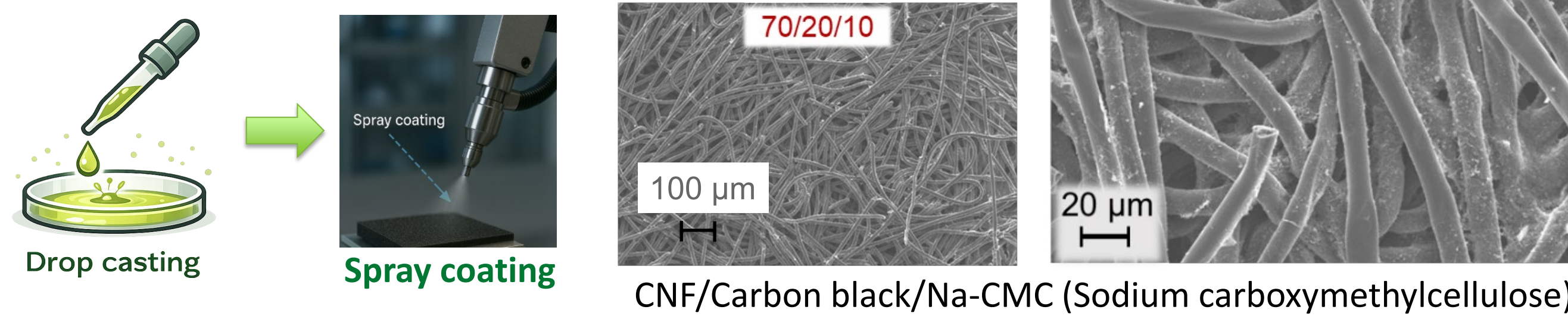
Solution: Develop **membrane-electrode assemblies (MEAs)** that enables controlled transport with cost-effective solutions

Approach: Address coupled bottlenecks through four coordinated strategies: (I) binder-stabilized carbon cathode hosts, (II) ionic-strength-balanced electrolytes, (III) PFAS-free membrane functionalization, and (IV) cation-diffusive carbon interlayers at the Na anode.

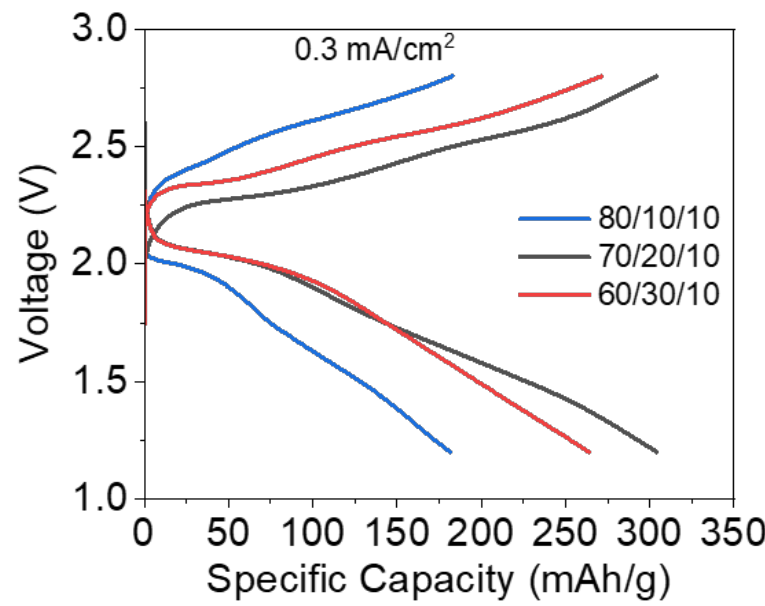


Results Card 1 - Catholyte utilization

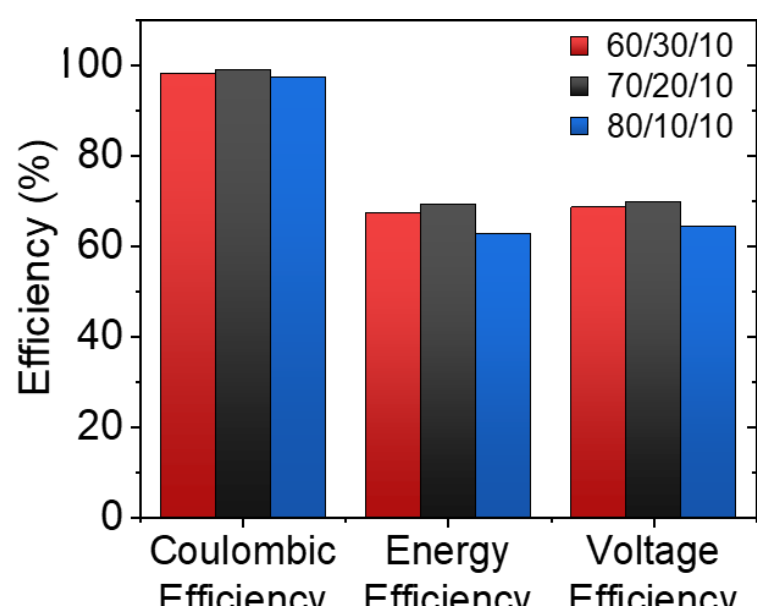
Design rule 1: Optimized CNF/CB/binder architecture increases sulfur utilization while preserving flow-compatible electrode integrity.



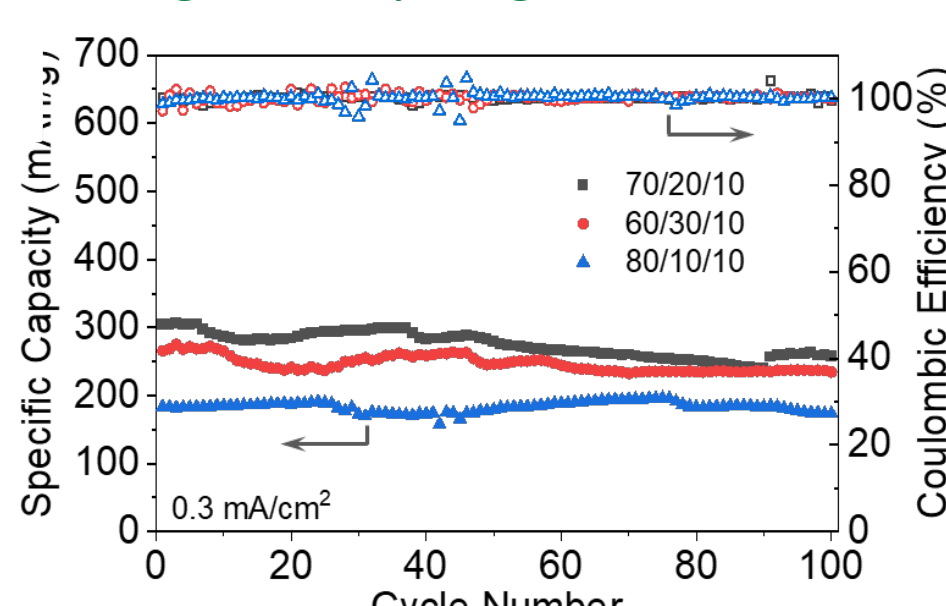
15th galvanostatic cycling profile



Comparison of efficiency



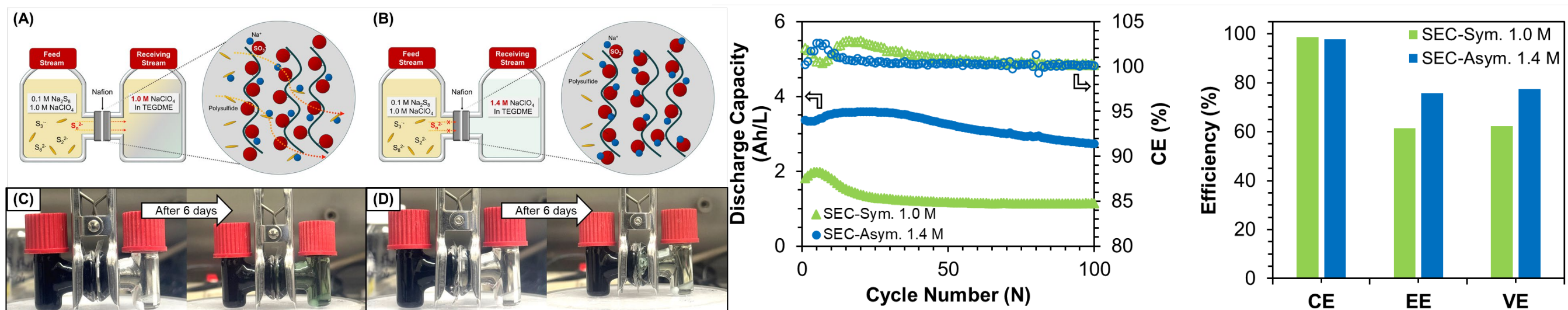
Long-term cycling



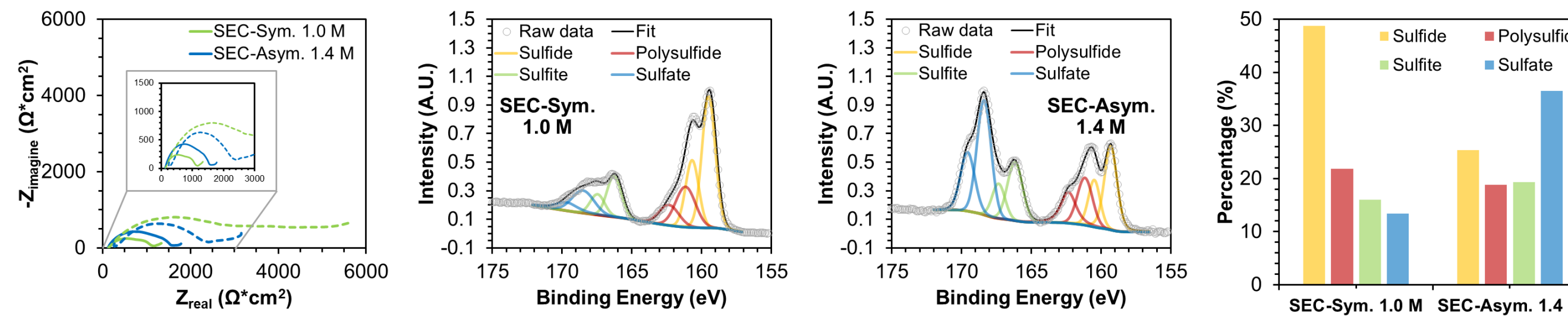
- Optimized 70/20/10 CNF/CB/Na-CMC composite on carbon paper gave the best balance of conductivity, binder distribution, porosity, and stability.
- **Mechanism:** CB improves redox kinetics; CNFs provide conductive pathways; Na-CMC helps stabilize the composite and improve cycling.

Results Card 2 - Electrolyte ionic-strength control

Design rule 2: Balancing anolyte/catholyte ionic strength suppresses polysulfide shuttling while preserving practical electrolyte conductivity.



- Polysulfide crossover reduced with anolyte contains higher salt concentrations;
- Full-cell demonstration showed better discharge capacity, energy/voltage efficiency, and improved capacity retention using the SEC-Asym. design.



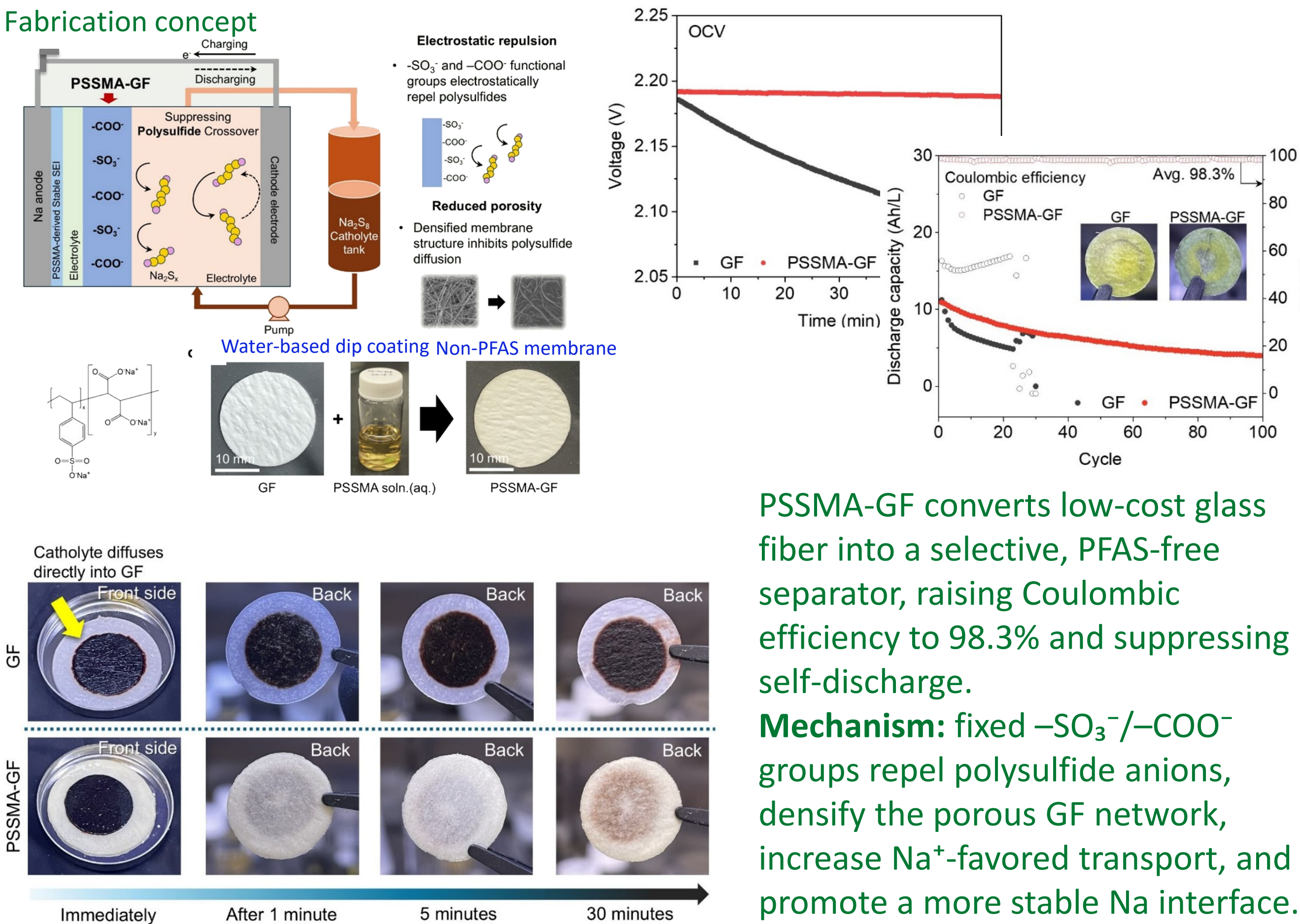
- Post-cell EIS and XPS confirm that the 1.4 M SEC-Asym electrolyte suppresses resistance growth and reduces polysulfide shuttling products from 70.61% to 44.19%.

Mechanism: Raise salt concentration in anolyte (governed by ionic-strength) reduces polysulfide shuttling, limits Na parasitic reactions, and preserves interfacial kinetics.

Results Card 3 - PFAS-free separator selectivity

Design rule 3: Oxyanion-functionalized porous separators electrostatically repel polysulfides while maintaining flow-compatible ion transport.

Fabrication concept



PSSMA-GF converts low-cost glass fiber into a selective, PFAS-free separator, raising Coulombic efficiency to 98.3% and suppressing self-discharge.

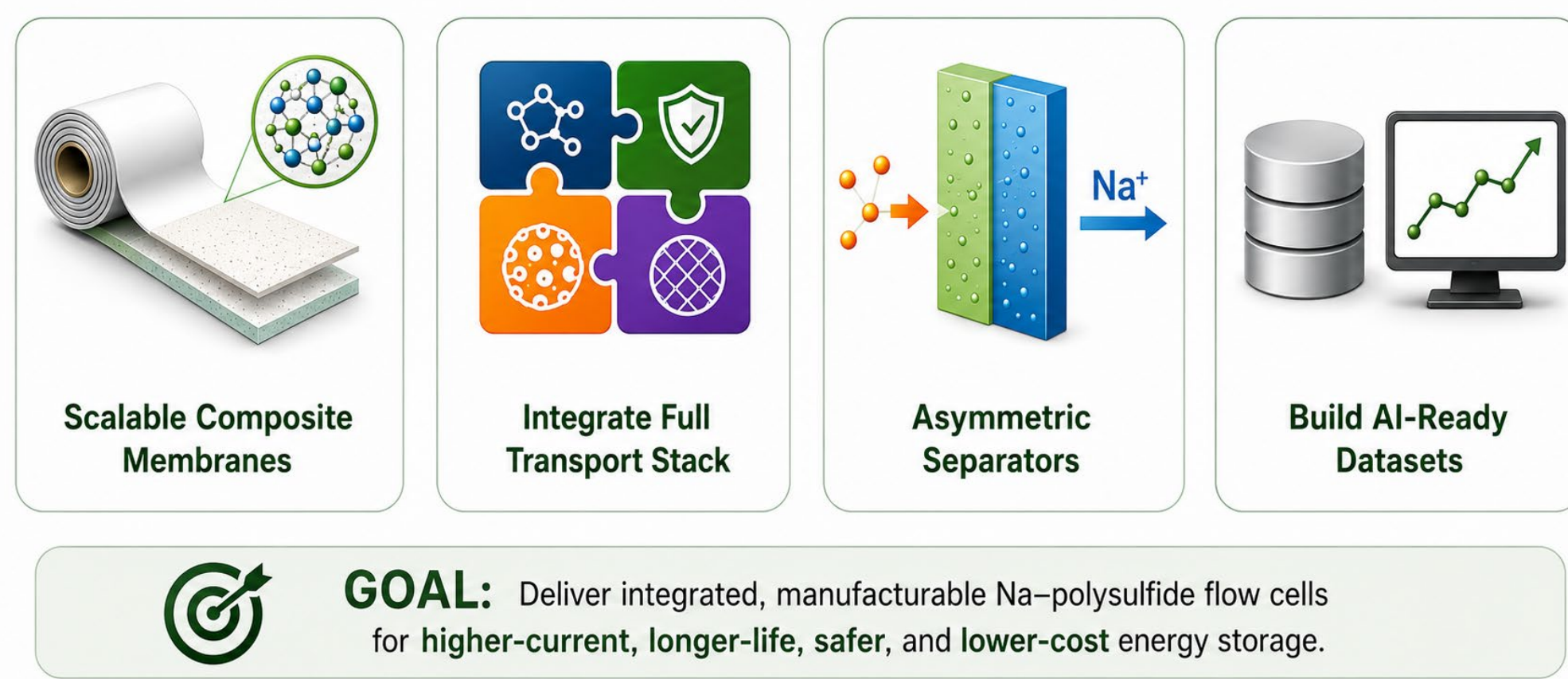
Mechanism: fixed -SO₃⁻/-COO⁻ groups repel polysulfide anions, densify the porous GF network, increase Na⁺-favored transport, and promote a more stable Na interface.

CONCLUSIONS & IMPACT

- Initiating an integrated transport-control strategy for Na–S flow batteries across multi-components;
- Improved material utilization, suppressed polysulfide crossover, reduced self-discharge, stabilized Na metal;
- Demonstrated scalable design routes using optimized carbon electrodes, electrolyte ionic-strength balancing, PFAS-free separator functionalization.

Impact: advances low-cost, earth-abundant Na–S redox-flow batteries toward durable long-duration energy storage for grid reliability and critical-load backup

FUTURE DIRECTIONS: Advancing Na–S Flow Batteries



FY26 ACHIEVEMENTS

Journal Publications (X5) [New since last OE review meeting]

Lead:

- Wenda Wu, Monojoy Goswami, Cheng-Tien Hsieh, Chanho Kim, Yuanshun Li, Nian Liu, and Guang Yang. "Cation-Diffusive Carbon Interlayers Stabilize Na Metal and Double the Current in Na-S Redox-Flow Batteries for Grid-Scale Energy Storage." *Adv. Funct. Mater.* **2026**, *36*, e31310.
- Jieun Kang, Cheng-Tien Hsieh, Wenda Wu, Guang Yang, and Nian Liu. "Suppressing Polysulfide Crossover in Sodium-Polysulfide Redox Flow Batteries with an Oxyanion-Functionalized Glass Fiber Separator." *ACS Nano* **2026**, *20*, 19401.
- Michelle L. Lehmann, Wenda Wu, Yuanshun Li, Chanho Kim, and Guang Yang. "Binder-Stabilized Carbon Fiber Hosts for High-Utilization Sodium–Polysulfide Flow Catholytes." *J. Electrochem. Soc.* **2026**, *173* 110512.
- Wenda Wu, Fuead Hasan, Yawei Gao, Chanho Kim, Inyoung Jang, Harry M. Meyer, Cheng Lei, Michelle Lehmann, Christopher M. Bejger, Xiaolin Li, and Guang Yang. "Suppressing Polysulfide Shuttling via a Supporting Electrolyte Concentration-Asymmetric (SEC-Asym) Design Based on Electrolyte Ionic Strength for Stable Promoted Na–Polysulfide Redox-Flow Batteries." *Energy Storage Mater.* **2026**, *90*, 105275.

Collaborated:

- Yutong Wu, Maxim Zhelyabovskiy, Zhitao Chen, Karam Eeso, Alexandros Filippas, Haochen Yang, Guang Yang, and Nian Liu. "Complexing Agent-Assisted Membraneless Zinc–Iodine Aqueous Batteries." *J. Phys. Chem. Lett.* **2025**, *16*, 8024.

Inventions (X1)

- Dry Processing of Ceramic–Polymer Hybrid Membranes Invention Disclosure ID# : 202506130, DOE S# (Inventors: Yang and Kim) [https://www.ornl.gov/technology/202506130]

ACKNOWLEDGMENTS

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